

## Chapter 22 Review Organic Chemistry Answers

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### Chapter 22 Review Organic Chemistry

Modern Chemistry 185 Organic Chemistry CHAPTER 22 REVIEW Organic Chemistry SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Match the reaction type on the left to its description on the right. \_\_\_\_ substitution \_\_\_\_ addition \_\_\_\_ condensation \_\_\_\_ elimination 2.

### CHAPTER 22 REVIEW Organic Chemistry

Loudon Chapter 22 Review: Enols and Enolates Jacquele Richardson, CU Boulder – Last updated 3/28/2019 2 α-Halogenation This just means removing an α H and attaching a halogen in its place. You can do this in either acid or base. In acid, you make the enol first, by the mechanism on the first page.

### Loudon Chapter 22 Review: Enols and ... - Organic Chemistry

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\\POLAND\_LAB\SYSTEMS\Teachers\BaSmith\Chemistry\O Chem\Introduction to Organic Chemistry.doc Major Understandings - 3.1 ff. gg, ii and 5.2 e Unit 7 Lesson - Introduction to Organic Chemistry Lesson 62 Pre-Assignment - None 1. Organic Compounds - any compound that contains the element Carbon. Actually oxides and carbonates aren't considered. ...

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Chapter 22 Review Questions: 1. How is organic chemistry different from inorganic chemistry? What type of molecules doe organic chemists study? a. Organic chemistry studies substances with carbon inorganic chemistry studies substances that do not contain carbon. 2. What are the most common elements from which the human body is made? a.

### Chapter 22 Review Questions - Chapter 22 Review Questions ...

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Organic Chemistry I Review: Highlights of Key Reactions, Mechanisms, and Principles 1 Some Arrow-Pushing Guidelines (Section 1.14) 1. Arrows follow electron movement. 2. Some rules for the appearance of arrows • The arrow must begin from the electron source. There are two sources: a. An atom (which must have a lone pair to give) b.

### Some Arrow-Pushing Guidelines (Section 1.14)

2. Which element is present in all organic compounds? Hydrogen Nitrogen Carbon Oxygen; 3. What products are obtained with CH 4 (g) burns completely in an excess of oxygen? CO and H 2 O CO and C CO 2 and H 2 O CO 2 and CO; 4. Which hydrocarbon is a member of the alkene series? C 2 H 2 C 3 H 6 C 4 H 10 C 4 H 12; 5. Which formula represents butane ...

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