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appropriate unit conversions have been applied.

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your hand at figuring out the amounts of reactants and products that take part in a chemical reaction. To begin, check that this equation is shown: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$.

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Activity A of the Stoichiometry Student Exploration. This investigation is to be used with the Stoichiometry Gizmo.

Questions 1 and 2: Assessment Citation:
Check Understanding Practice Skill

Students use the concept of a mole to answer basic questions related to chemical reactions. Activity A of the Stoichiometry Student Exploration.

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The Stoichiometry Gizmo™ allows you to try your hand at figuring out the

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amounts of reactants and products that take part in a chemical reaction. To begin, check that this equation is shown: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$ 1. Look at the coefficients (such as the “3” in 3CO) in front of each substance in the equation.

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$\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$ 1. Look at the coefficients (such as the “3” in 3CO) in front of each substance in the equation.

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