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The MoleThe Mole

280 Chemistry: Matter and Change • Chapter 14 Solutions Manual CHAPTER 14 SOLUTIONS MANUAL 32. Explain the similarities and differences

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between a 1 M solution of NaOH and a 1 m solution of NaOH. Both solutions contain NaOH (solute) dissolved in water (solvent). The 1 m solution contains 1 mole of NaOH per kilogram of water; the 1M solution 3 _ 5 ___ 1 _ _ _ _ _ (_

Mixtures and Solutions

Solutions Manual Chemistry: Matter and Change • Chapter 9 145 CHAPTER 9 SOLUTIONS MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide. chemical equation: $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$ complete ionic equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

Chemical Reactions

86 Chemistry: Matter and Change • Chapter 6 Solutions Manual CHAPTER 6 SOLUTIONS MANUAL 4. Determine why producing 1 million francium atoms per second is not enough to make measurements, such as density or boiling point. Even one million atoms collected together as a solid are microscopic. A grain of salt contains about 10^{15} sodium atoms.

The Periodic Table and Periodic Law

124 Chemistry: Matter and Change • Chapter 8 Solutions Manual 45. O 3 O O O O O 46. Challenge Draw the Lewis resonance structure for the ion SO_3^{2-} . O S O 2- O O O O O S 2- S 2- Draw the expanded octet Lewis structure of each molecule. 47. ClF_3 48. PCl_5 49. Challenge Draw the Lewis structure for the molecule formed when six fluorine atoms and

Covalent Bonding

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